

Molecules in Magnetic Fields

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Sections

- 1 Electronic Hamiltonian
- 2 London Orbitals
- 3 Paramagnetism and diamagnetism

Section 1

Electronic Hamiltonian

Hamiltonian Mechanics

- ▶ In classical **Hamiltonian mechanics**, a system of particles is described in terms their **positions** q_i and **conjugate momenta** p_i .
- ▶ For each system, there exists a scalar **Hamiltonian function** $H(q_i, p_i)$ such that the classical equations of motion are given by:

$$\dot{q}_i = \frac{\partial H}{\partial p_i}, \quad \dot{p}_i = -\frac{\partial H}{\partial q_i} \quad (\text{Hamilton's equations})$$

- ▶ note: the Hamiltonian H is not unique!
- ▶ Example: a single particle of mass m in a conservative force field $F(q)$
 - ▶ the Hamiltonian is constructed from the corresponding **scalar potential**:

$$H(q, p) = \frac{p^2}{2m} + V(q), \quad F(q) = -\frac{\partial V(q)}{\partial q}$$

- ▶ Hamilton's equations of motion are equivalent to Newton's equations:

$$\left. \begin{aligned} \dot{q} &= \frac{\partial H(q,p)}{\partial p} = \frac{p}{m} \\ \dot{p} &= -\frac{\partial H(q,p)}{\partial q} = -\frac{\partial V(q)}{\partial q} \end{aligned} \right\} \implies m\ddot{q} = F(q) \quad (\text{Newton's equations})$$

- ▶ Hamilton's equations are first-order differential equations – Newton's are second-order

Quantization of a Particle in a Conservative Force Field

- ▶ The Hamiltonian formulation is **more general** than the Newtonian formulation:
 - ▶ it is invariant to coordinate transformations
 - ▶ it provides a uniform description of matter and field
 - ▶ it constitutes the springboard to quantum mechanics
- ▶ The Hamiltonian function (total energy) of a particle in a **conservative force field**:

$$H(q, p) = \frac{p^2}{2m} + V(q)$$

- ▶ Standard rule for **quantization** (in Cartesian coordinates):
 - ▶ carry out the **operator substitutions**

$$\mathbf{p} \rightarrow -i\hbar\nabla, \quad H \rightarrow i\hbar\frac{\partial}{\partial t}$$

- ▶ multiply the resulting expression by the **wave function** $\Psi(q)$ from the right:

$$i\hbar\frac{\partial\Psi(q)}{\partial t} = \left[-\frac{\hbar^2}{2m}\nabla^2 + V(q) \right] \Psi(q)$$

- ▶ This approach is sufficient for a treatment of electrons in an **electrostatic field**
 - ▶ it is **insufficient for nonconservative systems**
 - ▶ it is therefore inappropriate for systems in a general electromagnetic field

Lorentz Force and Maxwell's Equations

- ▶ In the presence of an **electric field** \mathbf{E} and a **magnetic field** (magnetic induction) \mathbf{B} , a classical particle of charge z experiences the **Lorentz force**:

$$\mathbf{F} = z(\mathbf{E} + \mathbf{v} \times \mathbf{B})$$

- ▶ since this force depends on the velocity \mathbf{v} of the particle, it is not conservative
- ▶ The electric and magnetic fields $\mathbf{E}(\mathbf{r}, t)$ and $\mathbf{B}(\mathbf{r}, t)$ satisfy **Maxwell's equations** (1861–1868):

$$\nabla \cdot \mathbf{E} = \rho/\epsilon_0 \quad \leftarrow \text{Coulomb's law}$$

$$\nabla \times \mathbf{B} - \epsilon_0\mu_0 \partial\mathbf{E}/\partial t = \mu_0\mathbf{J} \quad \leftarrow \text{Ampère's law with Maxwell's correction}$$

$$\nabla \cdot \mathbf{B} = 0$$

$$\nabla \times \mathbf{E} + \partial\mathbf{B}/\partial t = \mathbf{0} \quad \leftarrow \text{Faraday's law of induction}$$

where $\rho(\mathbf{r}, t)$ and $\mathbf{J}(\mathbf{r}, t)$ are the **charge** and **current densities**, respectively

- ▶ Note:
 - ▶ when ρ and \mathbf{J} are known, Maxwell's equations can be solved for \mathbf{E} and \mathbf{B}
 - ▶ but the particles are driven by the Lorentz force, so ρ and \mathbf{J} are functions of \mathbf{E} and \mathbf{B}
- ▶ We here consider the **motion of particles in a given (fixed) electromagnetic field**

Scalar and Vector Potentials

- The second, **homogeneous pair of Maxwell's equations** involves only **E** and **B**:

$$\nabla \cdot \mathbf{B} = 0 \quad (1)$$

$$\nabla \times \mathbf{E} + \frac{\partial \mathbf{B}}{\partial t} = 0 \quad (2)$$

- ① Eq. (1) is satisfied by introducing the **vector potential A**:

$$\nabla \cdot \mathbf{B} = 0 \quad \Longrightarrow \quad \mathbf{B} = \nabla \times \mathbf{A} \quad \leftarrow \text{vector potential} \quad (3)$$

- ② inserting Eq. (3) in Eq. (2) and introducing a **scalar potential ϕ** , we obtain

$$\nabla \times \left(\mathbf{E} + \frac{\partial \mathbf{A}}{\partial t} \right) = 0 \quad \Longrightarrow \quad \mathbf{E} + \frac{\partial \mathbf{A}}{\partial t} = -\nabla \phi \quad \leftarrow \text{scalar potential}$$

- The second pair of Maxwell's equations is thus automatically satisfied by writing

$$\begin{aligned} \mathbf{E} &= -\nabla \phi - \frac{\partial \mathbf{A}}{\partial t} \\ \mathbf{B} &= \nabla \times \mathbf{A} \end{aligned}$$

- The potentials **(ϕ , **A**)** contain **four rather than six components** as in **(**E**, **B**)**.

- ϕ and **A** are obtained by solving the inhomogeneous pair of Maxwell's equations, containing ρ and **J**

Gauge Transformations

- ▶ Consider the following **gauge transformation** of the potentials:

$$\left. \begin{aligned} \phi' &= \phi - \frac{\partial f}{\partial t} \\ \mathbf{A}' &= \mathbf{A} + \nabla f \end{aligned} \right\} \text{with } f = f(\mathbf{q}, t) \leftarrow \text{gauge function of position and time}$$

- ▶ Such a transformation of the potentials does not affect the physical fields:

$$\begin{aligned} \mathbf{E}' &= -\nabla\phi' - \frac{\partial\mathbf{A}'}{\partial t} = -\nabla\phi + \nabla\frac{\partial f}{\partial t} - \frac{\partial\mathbf{A}}{\partial t} - \frac{\partial\nabla f}{\partial t} = \mathbf{E} \\ \mathbf{B}' &= \nabla \times \mathbf{A}' = \nabla \times (\mathbf{A} + \nabla f) = \mathbf{B} + \nabla \times \nabla f = \mathbf{B} \end{aligned}$$

- ▶ Conclusion: **the scalar and vector potentials ϕ and \mathbf{A} are not unique**
 - ▶ we are free to choose $f(\mathbf{q}, t)$ to make the potentials satisfy additional conditions
 - ▶ typically, we require the vector potential to be **divergenceless**:

$$\nabla \cdot \mathbf{A}' = 0 \implies \nabla \cdot (\mathbf{A} + \nabla f) = 0 \implies \nabla^2 f = -\nabla \cdot \mathbf{A} \leftarrow \text{Coulomb gauge}$$

- ▶ We shall always assume that the **vector potential satisfies the Coulomb gauge**:

$$\nabla \times \mathbf{A} = \mathbf{B}, \quad \nabla \cdot \mathbf{A} = 0 \quad \leftarrow \text{Coulomb gauge}$$

- ▶ note: \mathbf{A} is still not uniquely determined, the following transformation being allowed:

$$\mathbf{A}' = \mathbf{A} + \nabla f, \quad \nabla^2 f = 0$$

Hamiltonian in an Electromagnetic Field

- ▶ We must construct a Hamiltonian function such that Hamilton's equations are equivalent to Newton's equation with the Lorentz force:

$$\dot{q}_i = \frac{\partial H}{\partial p_i} \quad \& \quad \dot{p}_i = -\frac{\partial H}{\partial q_i} \quad \iff \quad m\mathbf{a} = z(\mathbf{E} + \mathbf{v} \times \mathbf{B})$$

- ▶ To this end, we introduce scalar and vector potentials ϕ and \mathbf{A} such that

$$\mathbf{E} = -\nabla\phi - \frac{\partial \mathbf{A}}{\partial t}, \quad \mathbf{B} = \nabla \times \mathbf{A}$$

- ▶ In terms of these potentials, the classical Hamiltonian function becomes

$$H = \frac{\pi^2}{2m} + z\phi, \quad \boldsymbol{\pi} = \mathbf{p} - z\mathbf{A} \quad \leftarrow \text{kinetic momentum}$$

- ▶ Quantization is then accomplished in the usual manner, by the substitutions

$$\mathbf{p} \rightarrow -i\hbar\nabla, \quad H \rightarrow i\hbar\frac{\partial}{\partial t}$$

- ▶ The time-dependent Schrödinger equation for a particle in an electromagnetic field:

$$i\hbar\frac{\partial\Psi}{\partial t} = \frac{1}{2m} (-i\hbar\nabla - z\mathbf{A}) \cdot (-i\hbar\nabla - z\mathbf{A})\Psi + z\phi\Psi$$

Electron Spin

- ▶ The **nonrelativistic Hamiltonian** for an electron in an electromagnetic field is then given by:

$$H = \frac{\pi^2}{2m} - e\phi, \quad \pi = -i\hbar\nabla + e\mathbf{A}$$

- ▶ However, this description ignores a fundamental property of the electron: **spin**.
- ▶ Spin was **introduced by Pauli in 1927**, to fit experimental observations:

$$H = \frac{(\boldsymbol{\sigma} \cdot \boldsymbol{\pi})^2}{2m} - e\phi = \frac{\pi^2}{2m} + \frac{e\hbar}{2m} \mathbf{B} \cdot \boldsymbol{\sigma} - e\phi$$

where $\boldsymbol{\sigma}$ contains three operators, represented by the **two-by-two Pauli spin matrices**

$$\sigma_x = \begin{pmatrix} 0 & 1 \\ 1 & 0 \end{pmatrix}, \quad \sigma_y = \begin{pmatrix} 0 & -i \\ i & 0 \end{pmatrix}, \quad \sigma_z = \begin{pmatrix} 1 & 0 \\ 0 & -1 \end{pmatrix}$$

- ▶ The Schrödinger equation now becomes a **two-component equation**:

$$\begin{pmatrix} \frac{\pi^2}{2m} - e\phi + \frac{e\hbar}{2m} B_z & \frac{e\hbar}{2m} (B_x - iB_y) \\ \frac{e\hbar}{2m} (B_x + iB_y) & \frac{\pi^2}{2m} - e\phi - \frac{e\hbar}{2m} B_z \end{pmatrix} \begin{pmatrix} \Psi_\alpha \\ \Psi_\beta \end{pmatrix} = E \begin{pmatrix} \Psi_\alpha \\ \Psi_\beta \end{pmatrix}$$

- ▶ Note: the two components are coupled **only in the presence of an external magnetic field**

Spin and Relativity

- ▶ The introduction of spin by Pauli in 1927 may appear somewhat ad hoc
- ▶ By contrast, spin arises **naturally from Dirac's relativistic treatment** in 1928
 - ▶ is spin a relativistic effect?
- ▶ However, **reduction of Dirac's equation to nonrelativistic form** yields the Hamiltonian

$$H = \frac{(\boldsymbol{\sigma} \cdot \boldsymbol{\pi})^2}{2m} - e\phi = \frac{\pi^2}{2m} + \frac{e\hbar}{2m} \mathbf{B} \cdot \boldsymbol{\sigma} - e\phi \neq \frac{\pi^2}{2m} - e\phi$$

- ▶ in this sense, spin is not a relativistic property of the electron
 - ▶ on the other hand, in the nonrelativistic limit, all magnetic fields disappear. . .
- ▶ We interpret $\boldsymbol{\sigma}$ by associating an **intrinsic angular momentum (spin)** with the electron:

$$\mathbf{s} = \hbar\boldsymbol{\sigma}/2$$

Molecular Electronic Hamiltonian

- ▶ The nonrelativistic Hamiltonian for an electron in an electromagnetic field is therefore

$$H = \frac{\pi^2}{2m} + \frac{e}{m} \mathbf{B} \cdot \mathbf{s} - e\phi, \quad \boldsymbol{\pi} = \mathbf{p} + e\mathbf{A}, \quad \mathbf{p} = -i\hbar\nabla$$

- ▶ expanding π^2 and assuming the Coulomb gauge $\nabla \cdot \mathbf{A} = 0$, we obtain

$$\begin{aligned} \pi^2 \Psi &= (\mathbf{p} + e\mathbf{A}) \cdot (\mathbf{p} + e\mathbf{A}) \Psi = p^2 \Psi + e\mathbf{p} \cdot \mathbf{A} \Psi + e\mathbf{A} \cdot \mathbf{p} \Psi + e^2 A^2 \Psi \\ &= p^2 \Psi + e(\mathbf{p} \cdot \mathbf{A}) \Psi + 2e\mathbf{A} \cdot \mathbf{p} \Psi + e^2 A^2 \Psi = (p^2 + 2e\mathbf{A} \cdot \mathbf{p} + e^2 A^2) \Psi \end{aligned}$$

- ▶ in molecules, the dominant electromagnetic contribution is from the nuclear charges:

$$\phi = -\frac{1}{4\pi\epsilon_0} \sum_K \frac{Z_K e}{r_K} + \phi_{\text{ext}}$$

- ▶ Summing over all electrons and adding pairwise Coulomb interactions, we obtain

$$\begin{aligned} H &= \sum_i \frac{1}{2m} p_i^2 - \frac{e^2}{4\pi\epsilon_0} \sum_{K_i} \frac{Z_K}{r_{iK}} + \frac{e^2}{4\pi\epsilon_0} \sum_{i>j} r_{ij}^{-1} && \leftarrow \text{zero-order Hamiltonian} \\ &+ \frac{e}{m} \sum_i \mathbf{A}_i \cdot \mathbf{p}_i + \frac{e}{m} \sum_i \mathbf{B}_i \cdot \mathbf{s}_i - e \sum_i \phi_i && \leftarrow \text{first-order Hamiltonian} \\ &+ \frac{e^2}{2m} \sum_i A_i^2 && \leftarrow \text{second-order Hamiltonian} \end{aligned}$$

Magnetic Perturbations

- ▶ In atomic units, the **molecular Hamiltonian** is given by

$$H = H_0 + \underbrace{\sum_i \mathbf{A}(\mathbf{r}_i) \cdot \mathbf{p}_i}_{\text{orbital paramagnetic}} + \underbrace{\sum_i \mathbf{B}(\mathbf{r}_i) \cdot \mathbf{s}_i}_{\text{spin paramagnetic}} - \sum_i \phi(\mathbf{r}_i) + \underbrace{\frac{1}{2} \sum_i A^2(\mathbf{r}_i)}_{\text{diamagnetic}}$$

- ▶ There are two kinds of magnetic perturbation operators:
 - ▶ the **paramagnetic operator** is linear and may lower or raise the energy
 - ▶ the **diamagnetic operator** is quadratic and always raises the energy
- ▶ There are two kinds of paramagnetic operators:
 - ▶ the **orbital paramagnetic operator** couples the field to the electron's orbital motion
 - ▶ the **spin paramagnetic operator** couples the field to the electron's spin
- ▶ In the study of magnetic properties, we are interested in two types of perturbations:
 - ▶ **uniform external magnetic field** \mathbf{B} , with vector potential

$$\mathbf{A}_{\text{ext}}(\mathbf{r}) = \frac{1}{2} \mathbf{B} \times \mathbf{r} \quad \text{leads to Zeeman interactions}$$

- ▶ **nuclear magnetic moments** \mathbf{M}_K , with vector potential

$$\mathbf{A}_{\text{nuc}}(\mathbf{r}) = \alpha^2 \sum_K \frac{\mathbf{M}_K \times \mathbf{r}_K}{r_K^3} \quad \text{leads to hyperfine interactions}$$

where $\alpha \approx 1/137$ is the **fine-structure constant**

Section 2

London Orbitals

Hamiltonian in a Uniform Magnetic Field

- ▶ The **nonrelativistic electronic Hamiltonian** (implied summation over electrons):

$$H = H_0 + \mathbf{A}(\mathbf{r}) \cdot \mathbf{p} + \mathbf{B}(\mathbf{r}) \cdot \mathbf{s} + \frac{1}{2} \mathbf{A}(\mathbf{r})^2$$

- ▶ The vector potential of the uniform field \mathbf{B} is given by:

$$\mathbf{B} = \nabla \times \mathbf{A} = \text{const} \implies \mathbf{A}_0(\mathbf{r}) = \frac{1}{2} \mathbf{B} \times (\mathbf{r} - \mathbf{O}) = \frac{1}{2} \mathbf{B} \times \mathbf{r}_0$$

- ▶ note: the **gauge origin \mathbf{O}** is **arbitrary!**
- ▶ The **orbital paramagnetic interaction** becomes:

$$\mathbf{A}_0(\mathbf{r}) \cdot \mathbf{p} = \frac{1}{2} \mathbf{B} \times (\mathbf{r} - \mathbf{O}) \cdot \mathbf{p} = \frac{1}{2} \mathbf{B} \cdot (\mathbf{r} - \mathbf{O}) \times \mathbf{p} = \frac{1}{2} \mathbf{B} \cdot \mathbf{L}_0$$

where we have introduced the **angular momentum relative to the gauge origin**:

$$\mathbf{L}_0 = \mathbf{r}_0 \times \mathbf{p}$$

- ▶ The **diamagnetic interaction** becomes:

$$\frac{1}{2} \mathbf{A}_0^2(\mathbf{r}) = \frac{1}{8} (\mathbf{B} \times \mathbf{r}_0) \cdot (\mathbf{B} \times \mathbf{r}_0) = \frac{1}{8} [B^2 r_0^2 - (\mathbf{B} \cdot \mathbf{r}_0)^2]$$

- ▶ The electronic Hamiltonian in a uniform magnetic field **depends on the gauge origin**:

$$H = H_0 + \frac{1}{2} \mathbf{B} \cdot \mathbf{L}_0 + \mathbf{B} \cdot \mathbf{s} + \frac{1}{8} [B^2 r_0^2 - (\mathbf{B} \cdot \mathbf{r}_0)^2]$$

- ▶ as we shall see, a change of the origin is a **gauge transformation**

Gauge Transformation of Schrödinger Equation

- ▶ What is the effect of a gauge transformation on the wave function?
- ▶ Consider a general **gauge transformation** for the electron (atomic units):

$$\mathbf{A}' = \mathbf{A} + \nabla f, \quad \phi' = \phi - \frac{\partial f}{\partial t}$$

- ▶ It can be shown that this represents a **unitary transformation** of $H - i\partial/\partial t$:

$$\left(H' - i\frac{\partial}{\partial t} \right) = \exp(-if) \left(H - i\frac{\partial}{\partial t} \right) \exp(if)$$

- ▶ In order that the Schrödinger equation is still satisfied

$$\left(H' - i\frac{\partial}{\partial t} \right) \Psi' \iff \left(H - i\frac{\partial}{\partial t} \right) \Psi,$$

the wave function undergoes a **compensating unitary transformation**:

$$\Psi' = \exp(-if) \Psi$$

- ▶ All **observable properties** such as the electron density are then **unaffected**:

$$\rho' = (\Psi')^* \Psi' = [\Psi \exp(-if)]^* [\exp(-if) \Psi] = \Psi^* \Psi = \rho$$

Gauge-Origin Transformations

- ▶ Different choices of **gauge origin** in the external vector potential

$$\mathbf{A}_0(\mathbf{r}) = \frac{1}{2} \mathbf{B} \times (\mathbf{r} - \mathbf{O})$$

are related by **gauge transformations**:

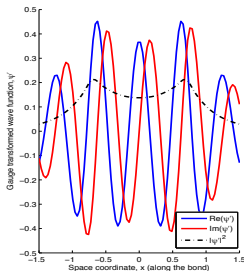
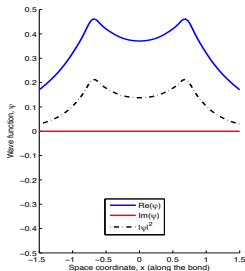
$$\mathbf{A}_G(\mathbf{r}) = \mathbf{A}_0(\mathbf{r}) - \mathbf{A}_0(\mathbf{G}) = \mathbf{A}_0(\mathbf{r}) + \nabla f, \quad f(\mathbf{r}) = -\mathbf{A}_0(\mathbf{G}) \cdot \mathbf{r}$$

- ▶ The exact wave function transforms accordingly and gives gauge-invariant results:

$$\psi_G^{\text{exact}} = \exp[-if(\mathbf{r})] \psi_0^{\text{exact}} = \exp[i\mathbf{A}_0(\mathbf{G}) \cdot \mathbf{r}] \psi_0^{\text{exact}} \quad (\text{rapid}) \text{ oscillations}$$

- ▶ **Illustration:** H_2 on the z axis in a magnetic field $B = 0.2$ a.u. in the y direction

- ▶ wave function with gauge origin at $\mathbf{O} = (0, 0, 0)$ (left) and $\mathbf{G} = (100, 0, 0)$ (right)



London Orbitals

- ▶ The **exact wave function** transforms in the following manner:

$$\Psi_{\mathbf{G}}^{\text{exact}} = \exp \left[i \frac{1}{2} \mathbf{B} \times (\mathbf{G} - \mathbf{O}) \cdot \mathbf{r} \right] \Psi_{\mathbf{O}}^{\text{exact}}$$

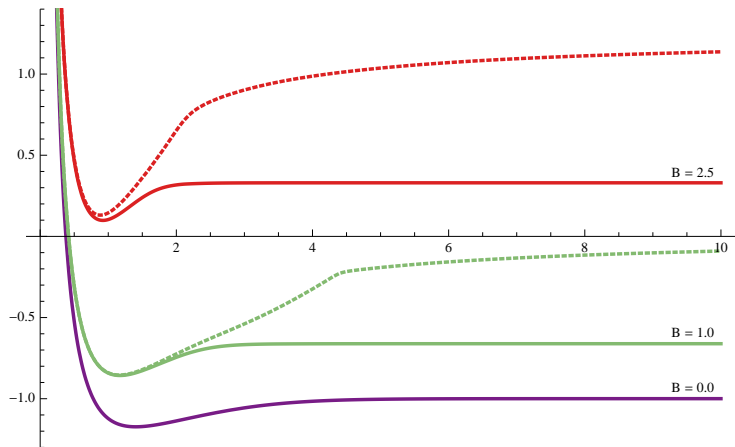
- ▶ this behaviour cannot easily be modelled by standard atomic orbitals
- ▶ Let us build this behaviour directly into the **atomic orbitals**:

$$\omega_{lm}(\mathbf{r}_{\mathbf{K}}, \mathbf{B}, \mathbf{G}) = \exp \left[i \frac{1}{2} \mathbf{B} \times (\mathbf{G} - \mathbf{K}) \cdot \mathbf{r} \right] \chi_{lm}(\mathbf{r}_{\mathbf{K}})$$

- ▶ $\chi_{lm}(\mathbf{r}_{\mathbf{K}})$ is a normal atomic orbital centred at \mathbf{K} and quantum numbers lm
- ▶ $\omega_{lm}(\mathbf{r}_{\mathbf{K}}, \mathbf{B}, \mathbf{G})$ is a field-dependent orbital at \mathbf{K} with field \mathbf{B} and gauge origin \mathbf{G}
- ▶ Each AO now responds in a **physically sound manner** to an applied magnetic field
 - ▶ indeed, all AOs are now **correct to first order** in \mathbf{B} , for any gauge origin \mathbf{G}
 - ▶ the calculations become rigorously gauge-origin independent
 - ▶ uniform (good) quality follows, independent of molecule size
- ▶ These are the **London orbitals** after Fritz London (1937)
 - ▶ also known as **GIAOs** (gauge-origin independent AOs or gauge-origin including AOs)
- ▶ Questions:
 - ▶ are London orbitals needed in atoms?
 - ▶ why not attach the phase factor to the total wave function instead?

Dissociation With and Without London Orbitals

- ▶ Let us consider the FCI dissociation of H_2 in a magnetic field
 - ▶ full lines: with London atomic orbitals
 - ▶ dashed lines: without London atomic orbitals



- ▶ Without London orbitals, the FCI method is not size extensive in magnetic fields

Section 3

Paramagnetism and diamagnetism

Paramagnetism

- ▶ Hamiltonian for a molecule in a **uniform magnetic field in the z direction**:

$$H = H_0 + \frac{1}{2}BL_z + Bs_z + \frac{1}{8}B^2(x^2 + y^2)$$

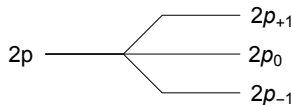
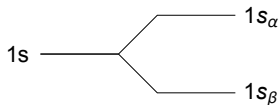
- ▶ a **paramagnetic, linear dependence** on the magnetic field
- ▶ a **diamagnetic, quadratic dependence** on the magnetic field
- ▶ The linear **paramagnetic Zeeman terms** are easily understood:
 - ▶ the angular momenta L_z and s_z set up a **magnetic moment**:

$$m_z = -\frac{1}{2}L_z - s_z$$

- ▶ this magnetic moment interacts with the field B in a **dipolar fashion**:

$$-Bm_z = \frac{1}{2}BL_z + Bs_z$$

- ▶ Important consequences of the **paramagnetic Zeeman terms**:
 - ▶ they **reduce symmetry and split energy levels**
 - ▶ energy is **raised or lowered**, depending on orientation



Diamagnetism

- ▶ Hamiltonian for a molecule in a **uniform magnetic field in the z direction**:

$$H = H_0 + \frac{1}{2}BL_z + Bs_z + \frac{1}{8}B^2(x^2 + y^2)$$

- ▶ The quadratic **diamagnetic term** may be understood in the following manner:

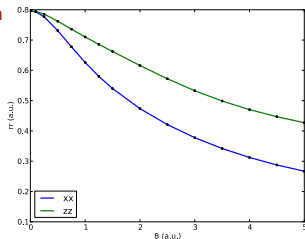
- 1 the field B induces a precession of the electrons with **Larmor frequency** $B/4\pi$
- 2 this precession generates an **induced magnetic moment** proportional to the field

$$\text{charge} \times \text{frequency} \times \text{area} = -\frac{B}{4\pi} \pi(x^2 + y^2)$$

- 3 this induced magnetic moment interacts with B , raising the energy quadratically

- ▶ Important consequences of **the diamagnetic term**

- 1 it **raises the energy of all systems**
- 2 it **squeezes all systems**
 - ▶ ground-state helium atom
 - ▶ **transversal size** $\propto 1/\sqrt{B}$
 - ▶ **longitudinal size** $\propto 1/\log B$

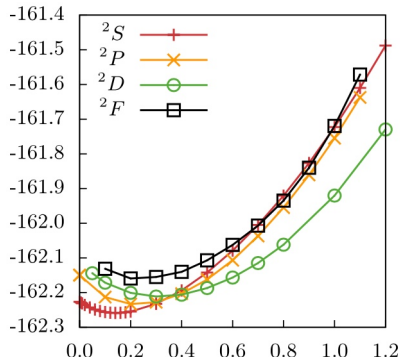
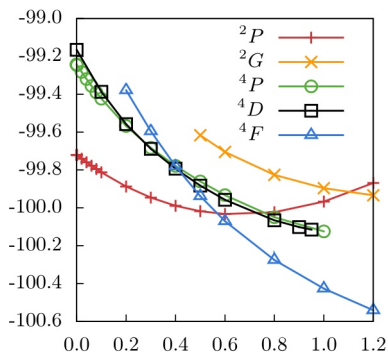


Open-shell systems – the quadratic Zeeman effect

- ▶ For open-shell atoms, we observe the **quadratic Zeeman effect**
 - ▶ initial energy lowering by Zeeman terms counteracted by the diamagnetic term

$$H = H_0 + Bs_z + \frac{1}{2}BL_z + \frac{1}{8}B^2(x^2 + y^2)$$

- ▶ Lowest states of the **fluorine atom** (left) and **sodium atom** (right) in a magnetic field



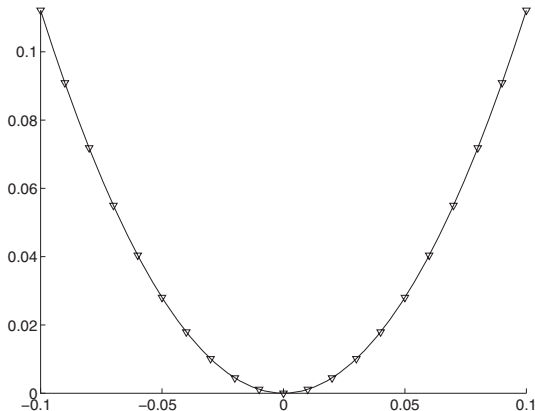
- ▶ CCSD(T) calculations in uncontracted aug-cc-pCVQZ basis (atomic units)

Closed-shell diamagnetism

- ▶ In a **closed-shell system**, ground-state energy should increase diamagnetically:

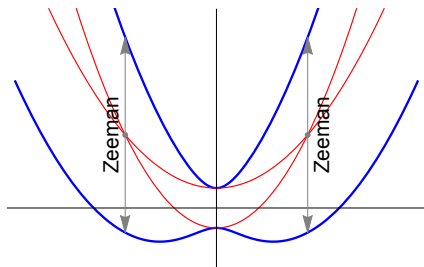
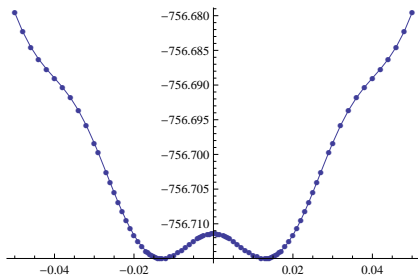
$$\langle 0|H|0\rangle = \langle 0|H_0|0\rangle + \frac{1}{8}B^2\langle 0|x^2 + y^2|0\rangle, \quad \langle 0|L_z|0\rangle = \langle 0|S_z|0\rangle = 0$$

- ▶ Energy of benzene in a perpendicular magnetic field (atomic units):



Closed-shell paramagnetism

- ▶ Nevertheless, closed-shell paramagnetic molecules such as C_{20} do exist



- ▶ Paramagnetism results from **Zeeman coupling of ground and excited states in the field**
 - ▶ in the absence of coupling, the diamagnetic **adiabatic ground and excited states** cross
 - ▶ the Zeeman interaction generates **adiabatic states** with an **avoided crossing**
 - ▶ a sufficiently strong coupling creates a **double minimum** (cmp. Renner–Teller)