

Something you didn't know: *all molecules are the same!*



September 13, 2017



Outline

- Stability of matter.
- Scaled energy/bond length relation, asymptotic behaviour.
- Numerical tests 1: behaviour with nuclear charge.
- Numerical tests 2: homologous series.
- Conclusions.



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- Original paper: NG, JPS, PRT, *Int J Quantum Chem*, **111**, 3324 (2011).



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Short r , large Z



ESQC, Torre Normanna, September 13, 2017

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- First answered by Dyson and Lenard in 1967 (following Onsager [classical particles] in 1939): affirmative **provided the fermion Pauli Principle is imposed**.
- Matter would not collapse without PP, but correct scaling with Z and N would not follow.
- Bosonic matter would follow a $-N^{7/5}$ law.



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Stability of the second kind

- “Dyson-Lenard proof one of the most difficult up to that time in the math phys literature” (Lieb).
- Simplified by Lieb and Thirring in 1975 using properties of Thomas-Fermi theory.
- Other issues arising from incorporation of relativity and of magnetic fields. Does not incorporate gravity (which of course *can* cause collapse).



Behaviour of the **exact** energy

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- Polyatomic case can be “collapsed” to the diatomic limit.

$$\begin{aligned}
 H &= -\frac{1}{2} \sum_i \nabla_i^2 + \sum_{i>j} r_{ij}^{-1} - \sum_{i,A} Z_A r_{iA}^{-1} + \sum_{A>B} Z_A Z_B R_{AB}^{-1} \\
 &\rightarrow -\frac{1}{2} \sum_i \nabla_i^2 + \sum_{i>j} r_{ij}^{-1} - \sum_{i,A} Z_A r_{iA}^{-1} + Z_A Z_B R_{AB}^{-1}
 \end{aligned}$$

but not further! (Must keep nuclear repulsion term.)



Behaviour of the **exact** energy

- Let the nuclear charges be Z_1e and Z_2e and let $Z = Z_1 + Z_2$. Total energy is E .
- Separated atom energies are E_1 and E_2 .
- Define the **scaled** internuclear distance $\rho = Z^{\frac{1}{3}}R$.
- Then (Lieb, Simon) as $Z \rightarrow \infty$, for finite ρ , all diatomic potential curves have the same leading term,

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- Leading term is in fact the **Thomas-Fermi (TF) energy**.



Thomas-Fermi energy

- Simple statistical model:

$$E_{TF} = E_{NR} + C \int [n(r)]^{5/3} dr + \int n(r) V_{NA}(r) dr + \frac{1}{2} \iint \frac{n(r)n(r')}{|r-r'|} dr dr'$$

$n(r)$ is the electron density, V_{NA} is the nuclear attraction, and E_{NR} is the nuclear repulsion energy.



Behaviour of the Hartree-Fock energy



Behaviour of the Hartree-Fock energy

- Exactly the same as the exact energy!
- Simply put, both the exact and Hartree-Fock energies have the same leading term — which is the TF energy — as $Z \rightarrow \infty$, for finite ρ .
- This includes $\rho = 0$.
- Also shown for a model with a relativistic kinetic energy operator and for next-higher terms in the energy expansion.



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- Not so!
- Consider for simplicity a homonuclear diatomic. Then $Z_1 Z_2 / R = 1/4\rho$ (certainly the same for all such diatomics).
- Define now $E_{\text{elec}} = E - Z_1 Z_2 / R$ and $\mathcal{E}_{\text{elec}} = Z^{-7/3} (E_{\text{elec}} - E_1 - E_2)$.
- Identical behaviour for different diatomics means identical $\mathcal{E}_{\text{elec}}$, not just that the latter is overwhelmed by the nuclear repulsion.



Not trivial, then, but good for what?

- Any known analytical behaviour is useful to calibrate and improve our methods.
- E.g., improving basis sets, or designing functionals in DFT.
- (Accepted, some analytical behaviour is useless, like asymptotic decay of density/orbitals in atoms.)



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Computational tests

- Can use Hartree-Fock since behaviour is the same as the exact energy.
- Not difficult to approach HF limit with modern basis sets.
- Must be prepared to approach/describe united-atom (UA) limit.
- Choose a variety of diatomics covering a range of Z (for simplicity, homonuclear diatomics with $^1\Sigma_g^+$ ground states), study R values inwards towards UA limit.



Computational methods

- Can't do much *ab initio* that's simpler than Hartree-Fock!
- Use correlation-consistent basis sets: cc-pCVQZ but only up to $l+2$ for the given atoms. (E.g., no g functions for N.)
- Large even-tempered *spd* or *spdf* set at bond midpoint. At small R a lot of linear dependence — canonical orthonormalization. No problems envisaged with BSSE this far up repulsive wall.
- Follow the lowest ${}^1\Sigma_g^+$ curve throughout (no idea if this remains the ground state for all $R \rightarrow 0$, of course).
- H_2 , He_2 , N_2 , Ne_2 , Ar_2 .



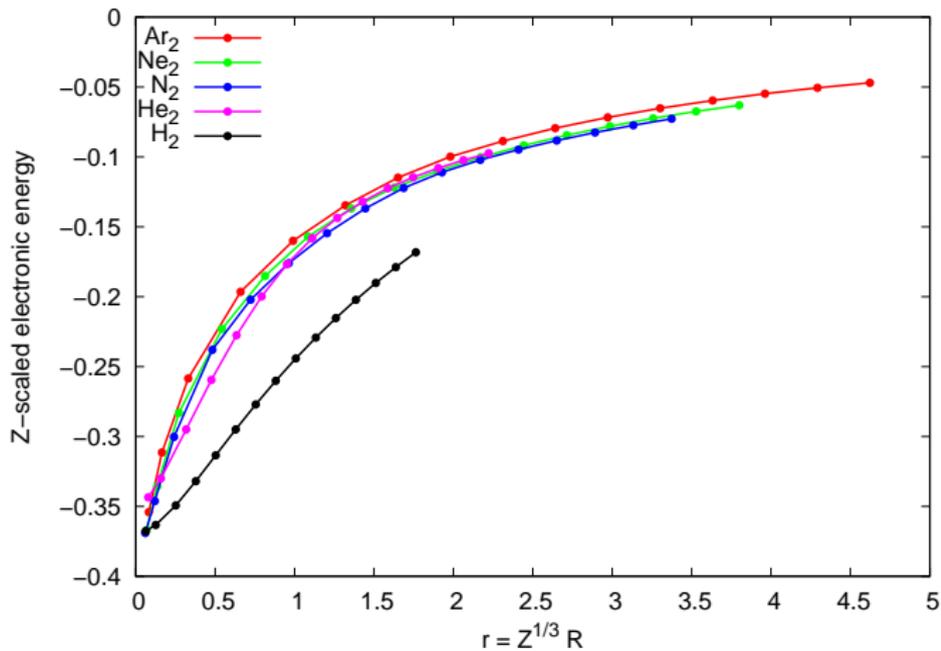
Computational methods

- Some potential wave function issues. HF UA for H_2 lowest $^1\Sigma_g^+$ state is He $1s^2$ ground state, but this is not true for UAs of the other diatomics.
- Use CASSCF with minimal active space needed for smooth transition to UA lowest “closed shell” (can be mixed like N_2 goes to Si ($^1D + ^1S$)).
- In practice see wave functions dominated by a single configuration in to short R , then rapid transition to UA single configuration. (He_2 transition to Be not observed at distances used.)

All calculations performed with Dalton.



Diatomic curves (a.u.)

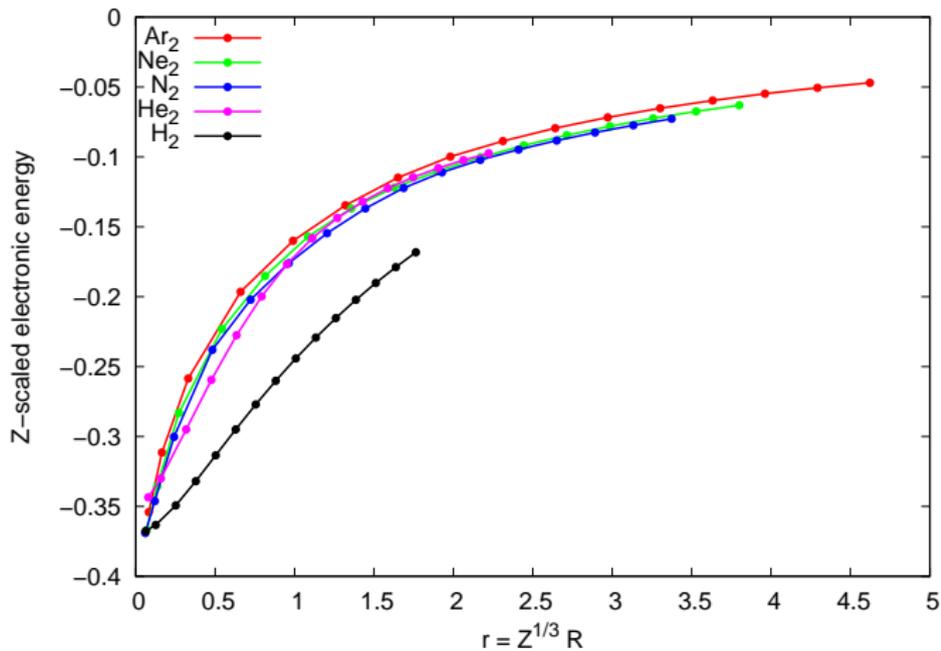


Distance comparisons (a_0)

ρ	H ₂ R	He ₂ R	N ₂ R	Ne ₂ R	Ar ₂ R
5.001			2.075		
1.765	1.4011				
1.5	1.191	0.945	0.622	0.553	0.454
1.0	0.794	0.630	0.415	0.368	0.303
0.5	0.397	0.315	0.207	0.184	0.151



Diatomic curves (a.u.)



Observations

- Not unexpectedly, H_2 is exceptional.
- He_2 behaves somewhat oddly at very short distances (possibly anticipating the transition to UA).
- The other three show essentially the same behaviour, basically to within mE_h , at all ρ values studied.
- Convergence to large Z behaviour appears exceptionally rapid, and is observed at relatively large ρ or R values.



Further tests

- The molecules examined are quite different in character: either strongly bonded, or not at all (in HF — very TF-like).
- Bound molecules that are “plausible” TF candidates?

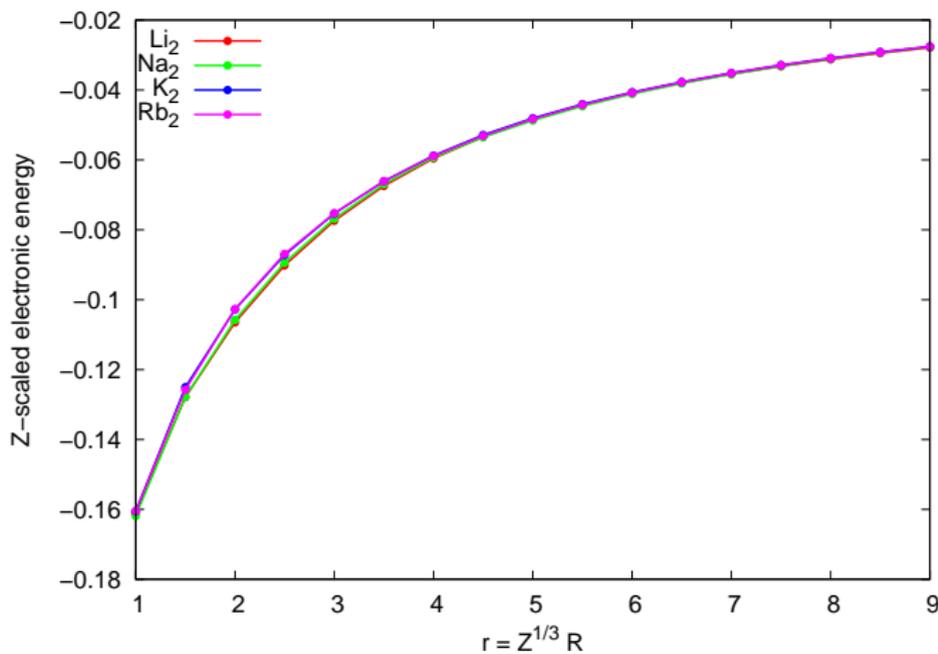


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- Bound molecules that are “plausible” TF candidates?
- Alkali-metal dimers. Noble gas-like cores inside a weak bonding pair.
- HF calculations on Li_2 , Na_2 , K_2 , Rb_2 . No bond functions, no CASSCF, as ultra-small R values not considered.



Alkali-metal dimers (a.u.)

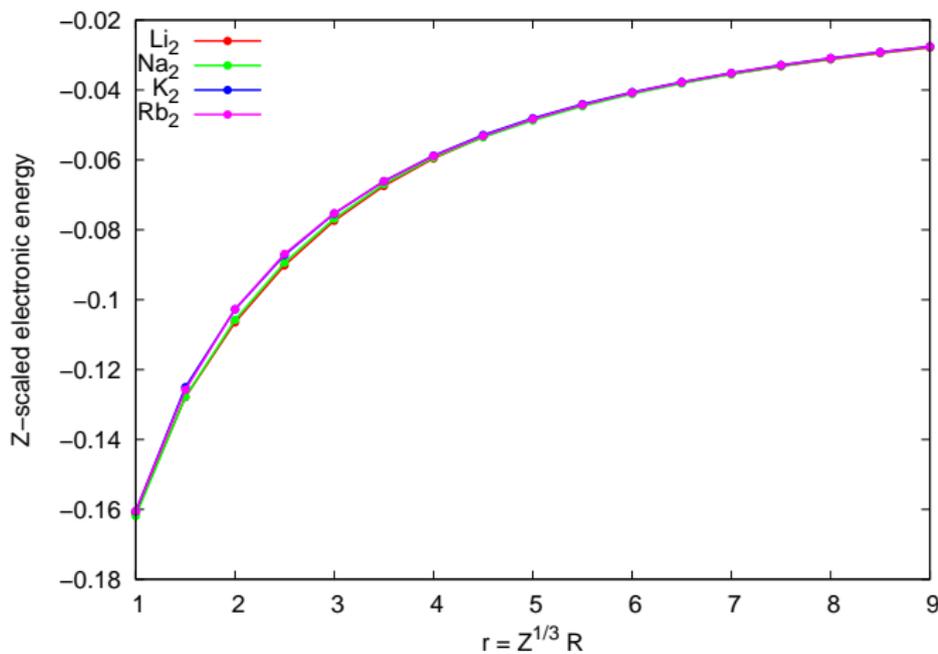


Distance comparisons (a_0)

	Li_2	Na_2	K_2	Rb_2
ρ	R	R	R	R
9.0	4.953	1.768	0.526	0.125
7.0	3.852	1.375	0.409	0.097
5.0	2.752	0.982	0.292	0.070
3.0	1.651	0.589	0.175	0.042



Alkali-metal dimers (a.u.)



Observations

- Again concordant to mE_h level, at relatively large ρ values (almost to R_e in $\text{Li}_2!$).
- Some divergence at ρ values between 2 and 3.
- Actual binding energies are rather small and badly underestimated by HF, which is partly responsible for similarity of curves for larger ρ : $\mathcal{E}_{\text{elec}}$ itself is becoming small.



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- Convergence to large Z behaviour is very rapid.
- Large Z behaviour observed at remarkably large ρ values.
- Note that the mathematics does not *exclude* that possibility: simply not proven (yet?).
- $\mathcal{E}_{\text{elec}}$ looks like a quantity that could be used to decide whether certain parametrized methods are realistic or not (proposed in first paper).
- Not really. In the intervening years: DFT, correlated methods, heteronuclear diatomics, open-shell systems —
all show the same behaviour.

